

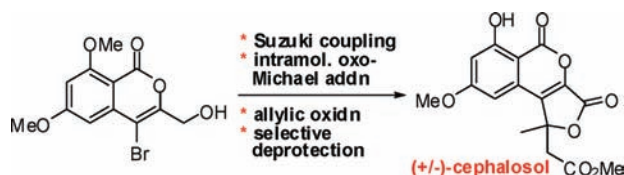
Total Synthesis of ( $\pm$ )-CephalosolYuanzhen Xie,<sup>†</sup> Ning Wang,<sup>‡</sup> Bin Cheng,<sup>‡</sup> and Hongbin Zhai<sup>\*,†,‡</sup>

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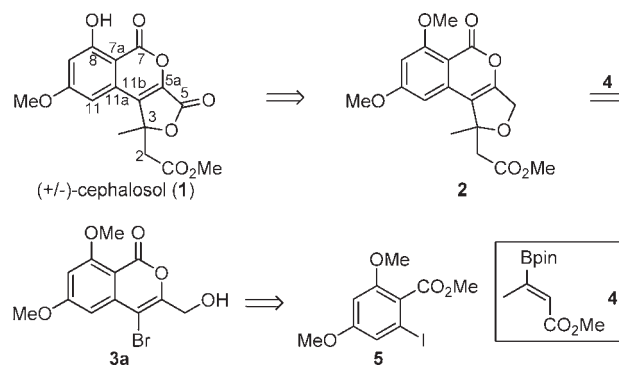
## ABSTRACT



A concise and efficient total synthesis of ( $\pm$ )-cephalosol has been completed (5 steps from known ester **5**, 39% overall yield), featuring a Cu(II)-promoted haloisocoumarin formation and sequential Suzuki cross-coupling/intramolecular oxo-Michael addition.

(–)-Cephalosol was isolated as a potent antimicrobial metabolite by Tan and co-workers from *Cephalosporium acremonium* IFB-E007, an endophytic fungal strain.<sup>1</sup> The host plant, *Trachelospermum jasminoides* (Lindl.) Lem. (Apocynaceae), has long been used in traditional Chinese medicine (TCM) to treat arthritis and other inflammatory diseases.<sup>1</sup> The natural product possesses a novel tricyclic backbone featuring a conjugated unsaturated  $\gamma$ -lactone fused to an isocoumarin at C-5a and C-11b. Attached to the sole quaternary center [C-3, of (*S*) configuration] are methyl and methoxycarbonylmethyl groups. Moreover, this metabolite showed prominent antimicrobial bioactivities as confirmed with human pathogenic microbes including *Escherichia coli*, *Pseudomonas fluorescens*, *Trichophyton rubrum*, and *Candida albicans*; the MIC values ranged from 1.95 to 7.8  $\mu\text{g}/\text{mL}$ . As a result, this molecule should be a superb target for the synthetic communities. Indeed, the first total synthesis of ( $\pm$ )-cephalosol (**1**, Scheme 1) has already been reported by Arlt and Koert.<sup>2</sup>

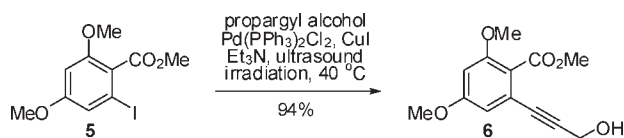
Herein, we wish to disclose our studies in developing a new convergent total synthesis. We envisioned that **1** could be constructed from **2** by allylic oxidation followed by a

Scheme 1. Retrosynthetic Analysis of ( $\pm$ )-Cephalosol (**1**)

selective ether cleavage. Tricycle **2** should be accessible via Suzuki coupling of bromoisocoumarin **3a** with boronate **4** followed by an intramolecular oxo-Michael addition. Bromoisocoumarin **3a** in turn could be generated from the known ester **5**<sup>3</sup> by Sonogashira coupling and Cu(II)-promoted haloisocoumarin formation.

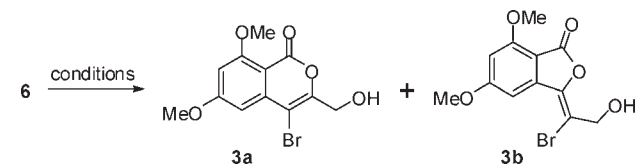
<sup>†</sup> University of Science and Technology of China.<sup>‡</sup> Lanzhou University.(1) Zhang, H. W.; Huang, W. Y.; Chen, J. R.; Yan, W. Z.; Xie, D. Q.; Tan, R. X. *Chem.—Eur. J.* **2008**, *14*, 10670.(2) Arlt, A.; Koert, U. *Synthesis* **2010**, 917 (requiring 10 steps).(3) Barros, M. T.; Maycock, C. D.; Madureira, M. I.; Ventura, M. R. *Chem. Commun.* **2001**, 37, 1662.(4) Takano, S.; Sugihara, T.; Samizu, K.; Akiyama, M.; Ogasawara, K. *Chem. Lett.* **1989**, *10*, 1781.(5) For representative examples, see: (a) Mehta, S.; Larock, R. C. *J. Org. Chem.* **2010**, *75*, 1652. (b) Roy, S.; Roy, S.; Neuenwander, B.; Hill, D.; Larock, R. C. *J. Comb. Chem.* **2009**, *11*, 1128. (c) Mehta, S.; Waldo, J. P.; Larock, R. C. *J. Org. Chem.* **2009**, *74*, 1141. (d) Chin, L. Y.; Lee, C. Y.; Lo, Y. H.; Wu, M. J. *J. Chin. Chem. Soc.* **2008**, *55*, 643. (e) Liang, Y.; Xie, Y. X.; Li, J. H. *Synthesis* **2007**, 400. (f) Yao, T.; Larock, R. C. *J. Org. Chem.* **2003**, *68*, 5936. (g) Yao, T.; Larock, R. C. *Tetrahedron Lett.* **2002**, *43*, 7401. (h) Biagetti, M.; Bellina, F.; Carpita, A.; Stabile, P.; Rossi, R. *Tetrahedron* **2002**, *58*, 5023.

## Scheme 2. Synthesis of Alkyne **6** via Sonogashira Coupling



Our synthesis commenced from ester **5**<sup>3</sup> (Scheme 2), which was smoothly converted into alkyne **6** by Sonogashira coupling with propargyl alcohol.<sup>4</sup> The synthesis of 4-haloisocoumarins have been extensively investigated during the past decade.<sup>5</sup> For instance,  $\text{Cy}_2\text{NH}\cdot\text{HX}$ <sup>5c</sup> could enhance  $\text{CuCl}_2$ - or  $\text{CuBr}_2$ -promoted cyclization of *o*-(1-alkynyl)benzoates for the synthesis of 4-haloisocoumarins. However, reaction of **6** by employing this protocol<sup>5c</sup> led to the formation of **3a** (6-*endo*, 13%) as the minor product compared to **3b** (5-*exo*, 70%, major), as shown in entry 1, Table 1. The cyclization conditions were optimized by scrutinizing the effects of different bases and temperatures (entries 2–6). To our delight, treatment of **6** with  $\text{CuBr}_2$  and pyridine in 1,2-dichloroethane at reflux produced **3a** and **3b** in 71% and 13% yields, respectively (entry 5).

Table 1. Optimization of  $\text{CuBr}_2$ -Promoted Cyclization of **6**<sup>a</sup>



entry	conditions	yield <sup>c</sup> ( <b>3a</b> / <b>3b</b> %)
1	$\text{CuBr}_2$ , $\text{Cy}_2\text{NH}\cdot\text{HBr}$ , 80 °C	13/70 <sup>b</sup>
2	$\text{CuBr}_2$ , reflux	28/38
3	$\text{CuBr}_2$ , $\text{K}_2\text{CO}_3$ , reflux	36/30
4	$\text{CuBr}_2$ , $\text{Cs}_2\text{CO}_3$ , reflux	42/28
5	<b><math>\text{CuBr}_2</math>, py, reflux</b>	<b>71/13</b>
6	$\text{CuBr}_2$ , DBU, reflux	10/6

<sup>a</sup> Reaction conditions: **6** (1.0 equiv),  $\text{CuBr}_2$  (2.1 equiv), base (1.5 equiv) in  $(\text{ClCH}_2)_2$  for 2 h. <sup>b</sup>  $\text{Cy}_2\text{NH}\cdot\text{HBr}$  (0.1 equiv). <sup>c</sup> Isolated yield.

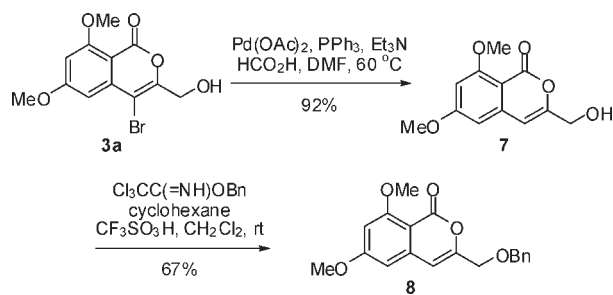
Since it was difficult to clearly distinguish the structures of **3a** and **3b** by <sup>1</sup>H and <sup>13</sup>C NMR analyses only, the presumed **3a** was transformed into the known compound **7**<sup>6</sup> via palladium-catalyzed debromination<sup>7</sup> with triethylammonium formate (Scheme 3). Unfortunately, our <sup>1</sup>H NMR data failed to match those documented for **7** in the literature.<sup>6</sup> Nevertheless, the data of the corresponding

(6) Saeed, A. *J. Heterocycl. Chem.* **2004**, *41*, 975.

(7) Cacchi, S.; Ciattini, P. G.; Morera, E.; Ortar, G. *Tetrahedron Lett.* **1986**, *27*, 5541.

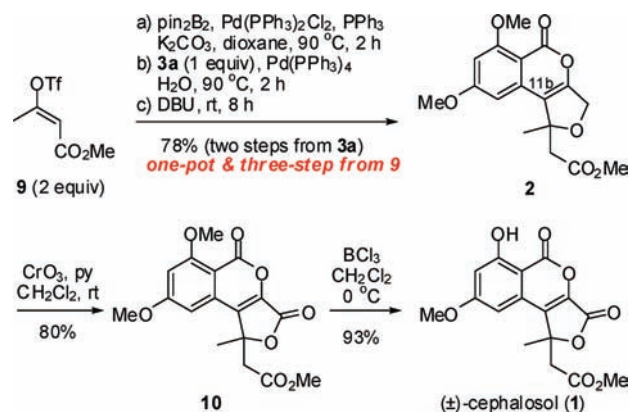
(8) Hager, A.; Mazunin, D.; Mayer, P.; Trauner, D. *Org. Lett.* **2011**, *13*, 1386.

## Scheme 3. Confirmation of the Structure of **3a**



benzyl ether **8**<sup>8</sup> (obtained by benzylation<sup>9</sup> of **7** with benzyl 2,2,2-trichloroacetimidate) were identical to those reported, which indicated that the NMR data for **7** in the literature<sup>6</sup> might not be accurate.

## Scheme 4. Completion of the Synthesis of (±)-Cephalosol (**1**)



With **3a** in hand, a four-carbon side chain had to be appended to C-11b<sup>10</sup> of **3a** in order to construct the lactone moiety (C ring) in **1**. Due to its susceptibility to protodeboronation,<sup>11</sup> boronate **4** must be generated in situ, for example, from enol triflate **9**<sup>12</sup> by Suzuki coupling with  $\text{pin}_2\text{B}_2$  (Scheme 4).<sup>13</sup> Without any purification, the freshly prepared boronate **4** was immediately exposed to **3a** in the presence of 10 mol % of  $\text{Pd}(\text{PPh}_3)_4$  in dioxane/ $\text{H}_2\text{O}$  (7:1) at 90 °C for 2 h. The reaction mixture was then directly treated with DBU to trigger the desired intramolecular oxo-Michael addition, and tricycle **2** was thus afforded in 78% yield. Finally, oxidation of **2** with the

(9) Keck, G. E.; Andrus, M. B.; Romer, D. R. *J. Org. Chem.* **1991**, *56*, 417.

(10) The numbering code used for cephalosol.

(11) Abraham, M. H.; Grellier, P. L. In *The Chemistry of the Metal-Carbon Bond*; Hartley, F. R., Patai, S., Eds.; Wiley: New York, 1985; Vol. 2, p 25.

(12) For the corresponding ethyl ester of (*E*)-**9**, see: Loreto, M. A.; Pompei, F.; Tardella, P. A.; Tofani, D. *Tetrahedron* **1997**, *53*, 15853. Compound (*Z*)-**9** is known according to a SciFinder search.

(13) (a) Takagi, J.; Takahashi, K.; Ishiyama, T.; Miyaura, N. *J. Am. Chem. Soc.* **2002**, *124*, 8001. (b) Ishiyama, T.; Takagi, J.; Kamon, A.; Miyaura, N. *J. Organomet. Chem.* **2003**, *687*, 284.

Collins reagent provided **10** (80%), which underwent selective ether cleavage at C-8 to give ( $\pm$ )-cephalosol (**1**) in 93% yield.<sup>2</sup>

In summary, we have accomplished a five-step total synthesis of ( $\pm$ )-cephalosol from ester **5** in 39% overall yield. Cu(II)-promoted haloisocoumarin formation and sequential Suzuki coupling/intramolecular oxo-Michael addition are worth noting for the current strategy.

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**Supporting Information Available.** Experimental procedures and analytical data of all new compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.